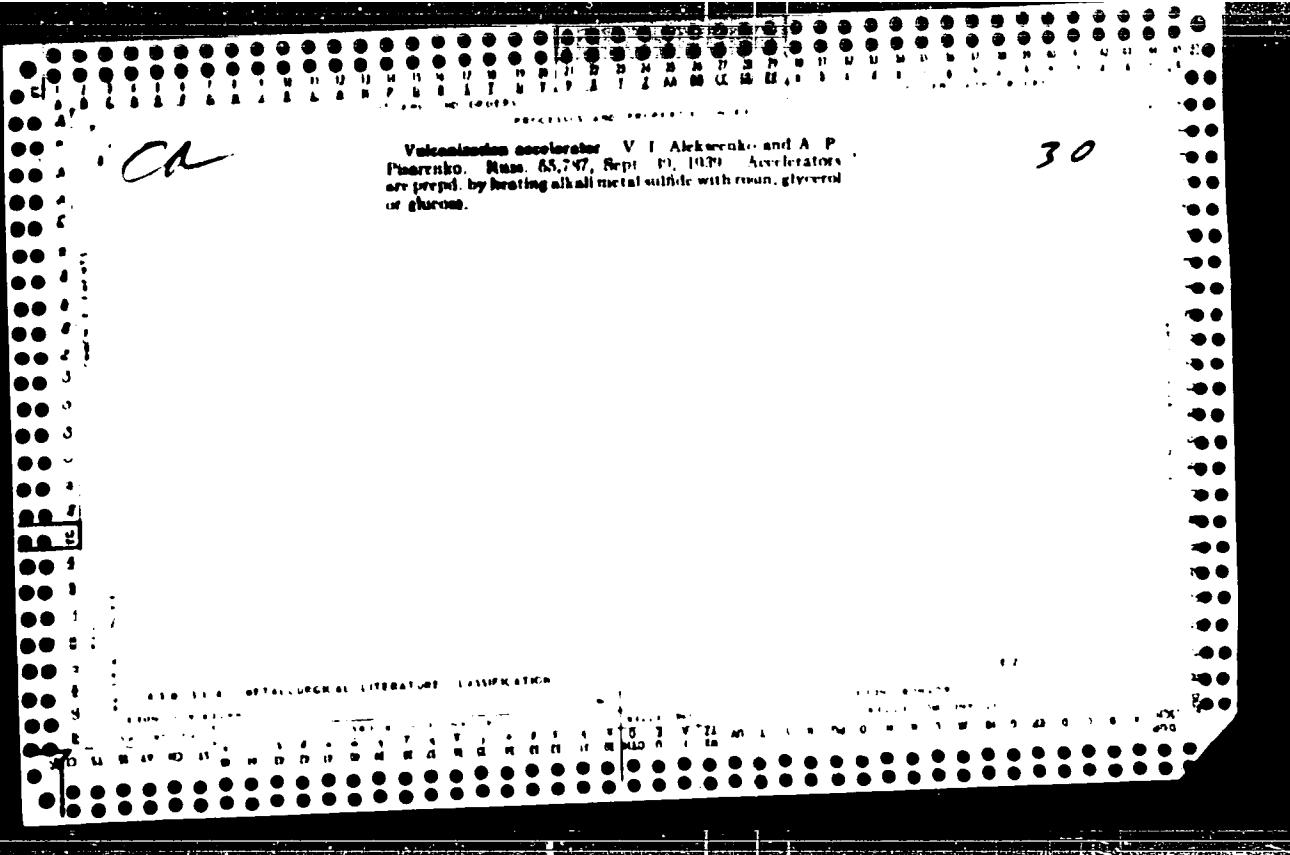


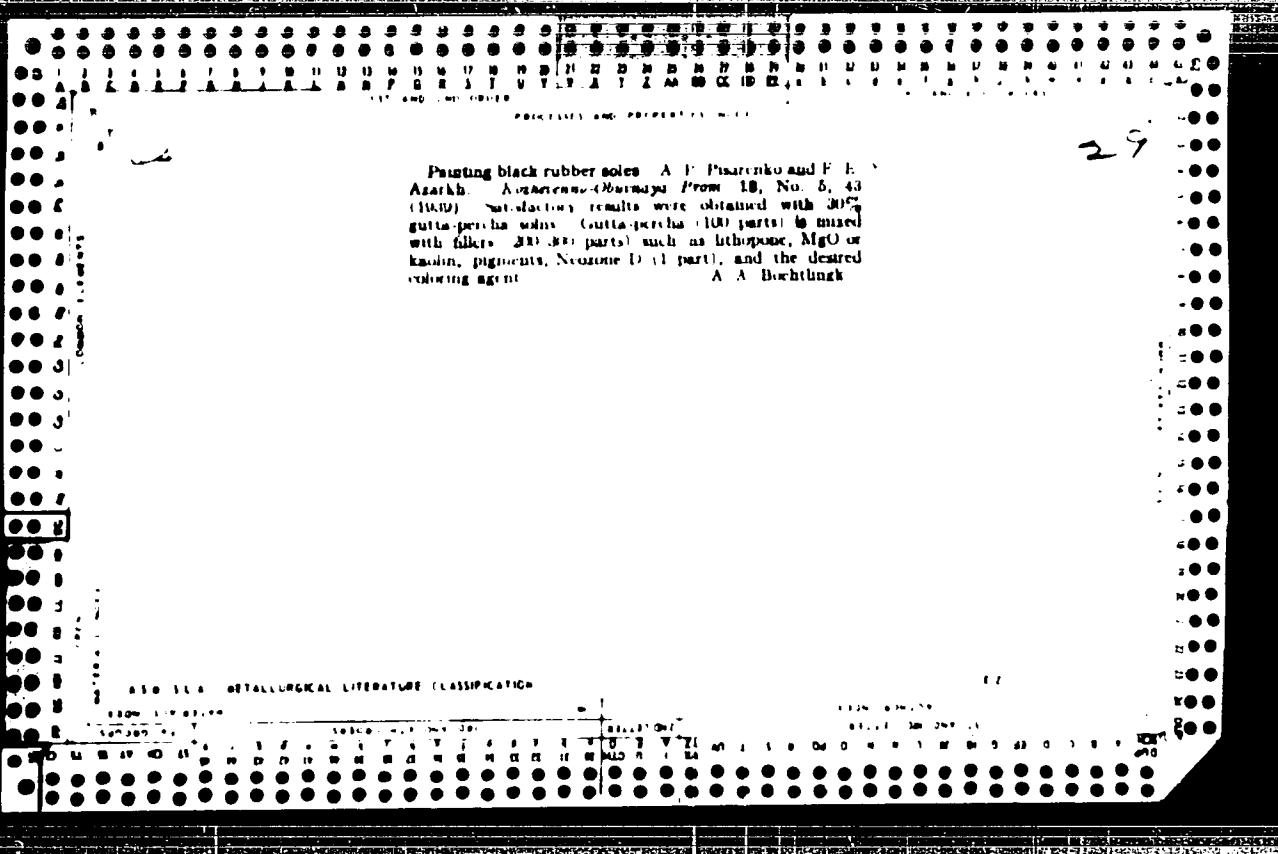
Production of colored reclaimed rubber from colored synthetic rubber vulcanizates A. P. Postnikov *Vestn. Chem. i Khim. Tekhn.* No. 3, 1961. The color and Rubber U.S.S.R. 1938, No. 13, 1961. The rubber was crushed (diam. of pieces 3-10 mm), mixed with 20-30% of (1) 80 parts of turpentine and 20 parts of paraffin or (2) 80 parts of turpentine and 20 parts of paraffin and the mixture was heated in a kettle for 4 hrs at 140° and the water was heated in a kettle for 4 hrs at 80° steam pressure. The reclaimed rubber with 5% S gave strengths of 54.5 kg per sq cm and 48.6 kg per sq cm for 25 and 40%, resp., of the 1st mixture of solvents above. A. Pestoff

Metal polysulfides as vulcanization agents and accelerators. A. I. Aleksenko and A. P. Pisarenko. Sov. Inventor's Certificate No. 17, No. 2, 1970, U.S.S.R. Inventor's Certificate No. 40, 941. The inventors tested many vulcanizing agents and the polysulfides. Among the polysulfide compounds, zinc polysulfide was found to give the shortest duration of vulcanization. The experiments show that the rate of vulcanization of Na-Cu-Zn polysulfide is much higher than that of vulcanization very considerably compared with that of mixes to which elemental S is added. A. P. S.

ASIN: A6-100-A6-100-B6-100-C6-100-D6-100-E6-100-F6-100-G6-100-H6-100-I6-100-J6-100-K6-100-L6-100-M6-100-N6-100-O6-100-P6-100-Q6-100-R6-100-S6-100-T6-100-U6-100-V6-100-W6-100-X6-100-Y6-100-Z6-100

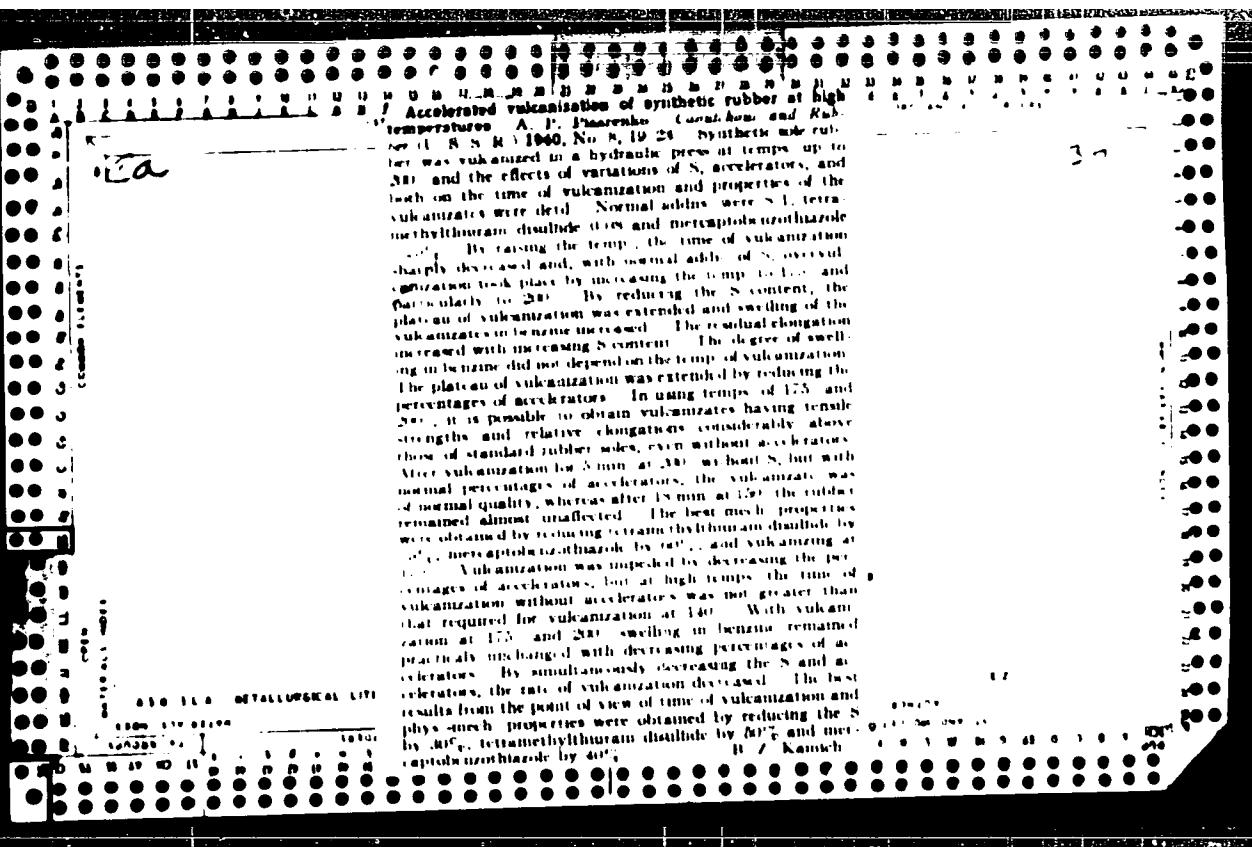
Disintegrating fiber fibers for the preparation of leather substitute. A. P. Pisarenko and M. Kh. Berishvili. Kakhovsko-Obninsk. Prom. 17, No. 4, 40-42 (1981). Chem. Zass. 1989, II, 303. Fibers 1 cm to 0.03 mm long were obtained after treating lunt with 1% HCl in a roller mill. The fibers were 0.03-0.014 mm long after treatment with a 5% HCl. The structure of the lunt remained unchanged after treatment with 5% HCl. The moisture content of the material is important for the disintegration of the fiber. The process of disintegration is shortened about 2-fold when the moisture content is lowered to 0.5%. A. A. Boehlitzk





The influence of the percentages of accelerators on the physical mechanical properties of colored rubber soles
L. U. Minchuk and A. P. Larenko. Koshvenno-
Obnensk Press, No. 9, 31(1959). Tests of the mech-
anical properties of soles made of Russian synthetic rubber with
various percentages of thiocaprolactam and tetra-
methylthiourea disulfide are reported. A. A. B.

29



Ca

The relation between physical-mechanical constants and the wear properties of rubber soles. A. A. Avilov and A. P. Parenko. *Izdatel. Nauk.-Tekhnichesk. Inst. Antarkticheskogo Obrabot. Promst.*, Sverdlovsk, Rabo 13, 107-311 (1940); *Chem. Zvez.* 1943, II, 304-7.—Tests were made to determine the relation between phys.-mech. properties and those observed in service. To evaluate the trib. characteristics of rubber sheets and the products made from them, the properties must be divided into 4 groups: (1) those that govern the strength of seams, (2) those that govern the resistance to abrasion and prevent defects not directly related to seams, (3) those that govern stability on aging and storage and (4) those that govern optimum processing. For (1), resistance to tear is responsible for durability in service. The values for new and cemented materials are somewhat different. For (2), resistance to friction is important. Values obtained with the Grasselli app. parallel practical experience. For (3), aging, especially artificial, lowers stretch appreciably, so limits can be defined as follows. The material is satisfactory if, after aging for 24 hrs at 70°, the extensibility is 16%—for (4), too great hard ness is not desirable; the optimum is 75-85 Shore D. —D. Clarke

3-2

Rapid vulcanization of auto rubber - A. V. Dzurok and N. P. Chelysheva. A review. Chemist. Prom. SSSR R-19 No. 3, p. 6 (1960). Vulcanization temps of 175 and 200° give vulcanizates with tensile strengths and relative elongations considerably above those required by standards for rubber soles. The elasticity of samples vulcanized at 175 or 200° was higher in all cases. By eliminating the accelerator and retaining a normal per centage of S, a normal vulcanizate is obtained in 3 min at 200°, whereas at 175° the rubber remains almost completely unvulcanized even after 18 min. Best results were obtained by lowering the thiuram compd by 30% and metapsophorothiazole by 10%, and vulcanizing at 175° Dtd No. 6, 20% It is possible to vulcanize black as well as colored shoe soles when sufficiently high temp are used. The amt of S and accelerators can be decreased considerably when the vulcanization temp is 225°. The higher vulcanization temps improve the quality of the goods, while a lowering in the S content increases the swelling in gasoline independently of the vulcanization temp. Aging of rubber is inhibited under the above conditions. A definite max temp is needed for each type of accelerator. Thus for the black rubber sole at 175° it is recommended to lower the S content by 10% of that of capra by 40% and of thiuram by 30%; while for 200° and 225° the decreases should be 15, 10 and 15%, resp.

A. A. Roehrling

Vulcanization of rubber with oxygen. A. D. Zinov'ev, N. V. Zinov'eva, A. P. Pavlenko, A. N. Zaonchikovskaya, S. D. Stevina, and N. N. Kostyleva. *Khimiia i Tekhnika Plasticheskikh Mass*, No. 8, p. 12, 1960. Catalyst begins to decompose at 150° and is a vulcanizing agent at 180°. With a product having a tensile strength of 60 kg/cm² at 150° and an elongation at rupture of 200-220%, was obtained benzoyl peroxide as the most active agent. With 1% vulcanization starts at 150°, with 1% at 200°. Nach's law of heat stability, promotes vulcanization only beginning with 200°. Bulk is active only above 200°. The experiments described and the results calculated and plotted.

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CA

Rubber Filler U.S. Mishustin, A. P. Psarenko, I. I.
Charkov, and M. Kh. Bernshtejn. U.S. Pat. No. 3,612,207
Date: Oct. 28, 1971. A filler for rubber is used lignin treated
with surface active substances having polar groups such
as dyes or soaps. U.S. Pat. No. 3,612,208, Oct. 28, 1971. Speci-
fies a filler of peat treated with hot water and solution of an
alkali hydroxide or NaSCN followed by acidification and
ground.

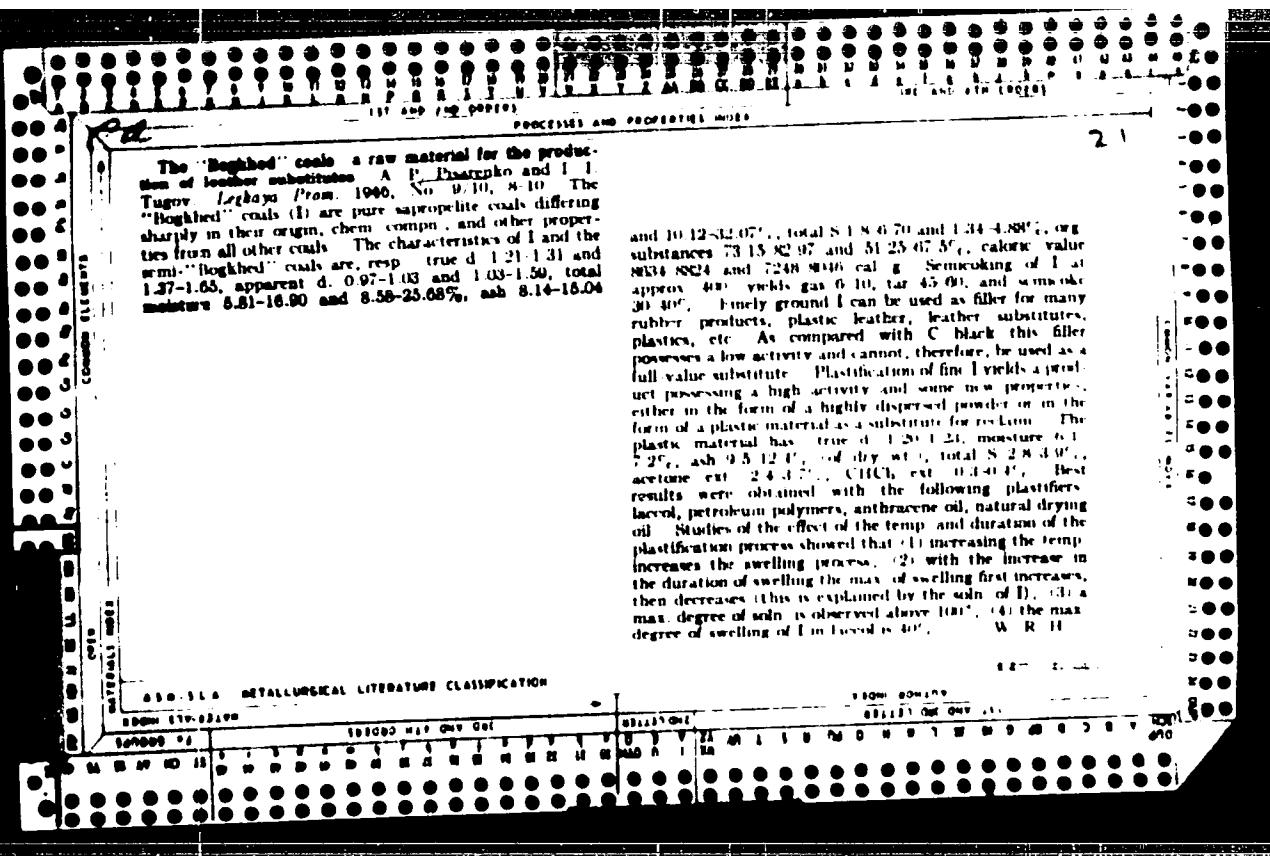
CF

Rubberlike Product A. G. Duff, A. M. L. T. R.
K. S. and A. D. Zelenchuk and U.S.S.R. 66-709
July 1, 1946. The invention relates to a process for
and certain substances are polymerized with rubber by
using in the presence of U.S.S.R. Rubber catalyst.

36

C

Activation of chalk, kaolin and other mineral fillers
for rubber. J. I. Tager and A. P. Pisarcikoff. U.S.P. No.
2,607,111, July 11, 1946. Mineral fillers for rubber com-
pounds impregnated with unsaturated, readily polymerizable, con-
jugated, or non-conjugated, polymeric substances.



New types of substitute sole leather A. P. Pustovets
and A. P. Kotel'yanova. *Izobraz. Prom.* 7, No. 1, 13-2
(1947). Substitutes for sole leathers were produced by impregnating waste fibers with various rubbery materials containing petroleum polymers. "Taklon" is a resin of polymerized unsat'd hydrocarbons in solvent naphtha; an oil-sol'd vegetable oils, resin, bitumen, etc. A resin of 100 parts of "Taklon", 8 parts of a steatite (Phacetate, Monostearate, etc.), and 10 parts of a rubbery resin esterified with Marshall's Styr.

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Theory of vulcanization of synthetic rubbers. A. P. Pisarenko and P. A. Reznikov. *Vulk. Prom.* 10, No. 9, 31 (1959); *J. Polym. Sci.*, Review of literature and 31 (1959), p. 1-44, 1959a. Review of literature and some exptl. work. Effect of vulcanization can be obtained by (1) difficultly destructible "rigid" internal bonds of the S bridge type, (2) elastic, and to a considerable degree reversible polar bonds. Addn. to the vulcanizates of reversible polar bonds. Addn. to the polar groupings (oxygen and substances which activate the polar groupings) facilitates the reversibility of the process. B. Z. K.

The vulcanization of synthetic rubbers A. P. ¹⁹⁵⁴
Lukin and P. A. Reznikov. Inst. Phys. Chem. Acad. Sci. S.S.R.
S.S.R. Moscow. Doklady Akad. Nauk S.S.R. 73, 120-122 (1950). Only a small fraction of the total S found in the process of vulcanization (7.1% in natural and 2.0% in synthetic rubber) forms spacial bridges between the macromolecules. On the other hand, vulcanizates can be obtained with as little as 0.10% bound S relative to the wt. of the rubber. The role of the bridge S in vulcanization appears entirely secondary. The main factor of the bound S is due by its adding to double bonds and formation of polar groups. The high cohesive forces of vulcanized rubbers are due to attraction between such polar groups. This point of view is corroborated by the decrease with increasing S of the initial strength of vulcanized rubbers illustrated by curves for acrylonitrile, Na-butadiene, and polychloroprene rubber. Vulcanization effects are observed under conditions where bridge formation by bound S is still out of the question. Swelling in Cellophane resulted in a regular fall of the strength and elongation on rupture of vulcanizates very nearly proportional to the sum of absorbed Cellophane and fully reversible on elimination of the Cellophane. This effect is analogous to that observed in unsaturable plastics of the type of polyvinyl acetate and chloride or polyvinylidene chloride where internal bonding is due to polar groups. In the vulcanization of Na-butadiene rubber with S and mixt. of C black decreases the fall of the degree of insulation and favors reversal of the vulcanization; this is an indication that vulcanization is largely a matter of adding to double bonds.

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BTR

9372 Dependence of Structure Forming Ability of Poly
mers on Method of Their Introduction Into Rubber To Be
Used As A Filler In Synthetic Rubber From Phenol-NaOH
Mixture. By V. V. Kostyuk, V. V. Gerasimov, V. V. Cherednik
and V. V. Slobodchikov. Academy of Sciences of the Ukrainian SSR

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Properties of vulcanized rubber as a function of tempera
ture A. P. Pisarenko Izdatelstvo Akademii Nauk SSSR
Review of contributions by Soviet scientists B. Z. K.

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UA687 - Vulcanization of Synthetic Rubbers A. I. Lissatnikov
and I. A. Kostyuk. *Rubber Chemistry and Technology*, Vol. 32,
No. 9, Sept. 1959, p. 569-574. Translated from *Doklady Akademii
Nauk SSSR*, Reports of the Academy of Sciences of the USSR,
Vol. 139, No. 1, 1961, p. 129-132.

This document is distributed from original.

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CA

Electron-microscopic investigation of ladder structures
A. P. Zhurenko, A. B. Shchukter, and A. I. Lebedeva [Inst.
Phys.-Chem. Acad. Sci. U.S.S.R., Moscow]. Doklady
Akad. Nauk SSSR 76, 423 (1951). Filmed 200-400 A
thickness 1.29 at 0.14%, double bonds in al.
per cent by weight of a toluene solution show characteristic
ridges oriented at different angles and occasionally forming
closed cells. With C black (II) filter and without a sur-
face active agent, the distribution of II is highly hetero-
geneous. 0.1% of stearic acid (III), benzidine (IV), or di-
phenylguanidine (V) does not change this picture, but with
1% III the dispersion of II is substantially improved, and
the dispersion of II is still further improved, but the no. of
the ridges decreases sharply. This is taken as an indication of
the purifying action of III on the gel fraction of I. IV and V
not only favor more homogeneous distribution of II, but in
crease the no. of ridges where most of the II is concen-
trated. Dispersion of whiting is not promoted by these surface ac-
tive substances.

PROKOP'YEV, Ya.N.; YEMEL'YANOVA, A.P.; PISARENKO, A.P.

Using high-styrene rubbers in the manufacture of microporous soles
for shoes. Kozh.-obuv. prom. no.3:19-23 Mr '59.

(MIRA 12:6)

(Rubber, Synthetic) (Shoe manufacture)

SCV/12-1-111/1

(..)
AUTHORS: Kudubov, Yu.F., Litvinova, A.I., Zubina, S.I., Shtukin,
B.V.

TITLE: Lignin as a Reinforcing Agent of SIS-IIC Rubber

PERIODICAL: Kolloidnyy Zhurnal, 1961, Vol XXI, No 1, p 107-110
(USSR)

ABSTRACT: The authors report on a study of the properties of vulcanizates, each containing a variety of activated lignin. The lignin varieties used for the experiments were powdered and ground with the vibromill, chlorinated and alkali-activated. Vulcanizates containing alkali-activated lignin were slightly reinforced. Electron microscopic investigation revealed that on activation of alkali-activated lignin appears in the formation of a reticular structure, which is similar to the structure of carbon black. This structure, which can not be seen in the other varieties, evidently accounts for the reinforcing effect of alkali-activated lignin. The authors mention the Soviet scientist N.N. Shery-

Card 1/2

PISARENKO, D.P.

✓ 018. Adhesion of high polymers. II. Developing a method for determining the adhesion of high polymers to one another. A. I. SHIROVALOVA, S. N. VOYERKIN, and A. P. PRYBILSKY. Koll. Zhur., 1950, 18, 483-63. In Russian, with English summary. A method has been developed for determining the mutual adhesion of two high polymers based on the pulling apart of their interface. The effect on the determination of the adhesion of butadieno-acrylonitrile polymer to Cellophane was ascertained of the ratio of the separation, the width of the interface, the thickness of the adhesive layer and the pressure exerted on the joint prior to the measurement. Heat treatment of the joints was found to considerably increase the adhesion. On lengthening the time of contact between the butadieno-acrylonitrile polymer and the Cellophane the adhesion increases, tending toward a given limit. It is concluded that the adhesion of a high polymer adhesive to a substrate, also of a high polymer, is due to the diffusion of the molecules of the former into the latter and vice versa. 3S2D31MN21.60513

LPH RTR

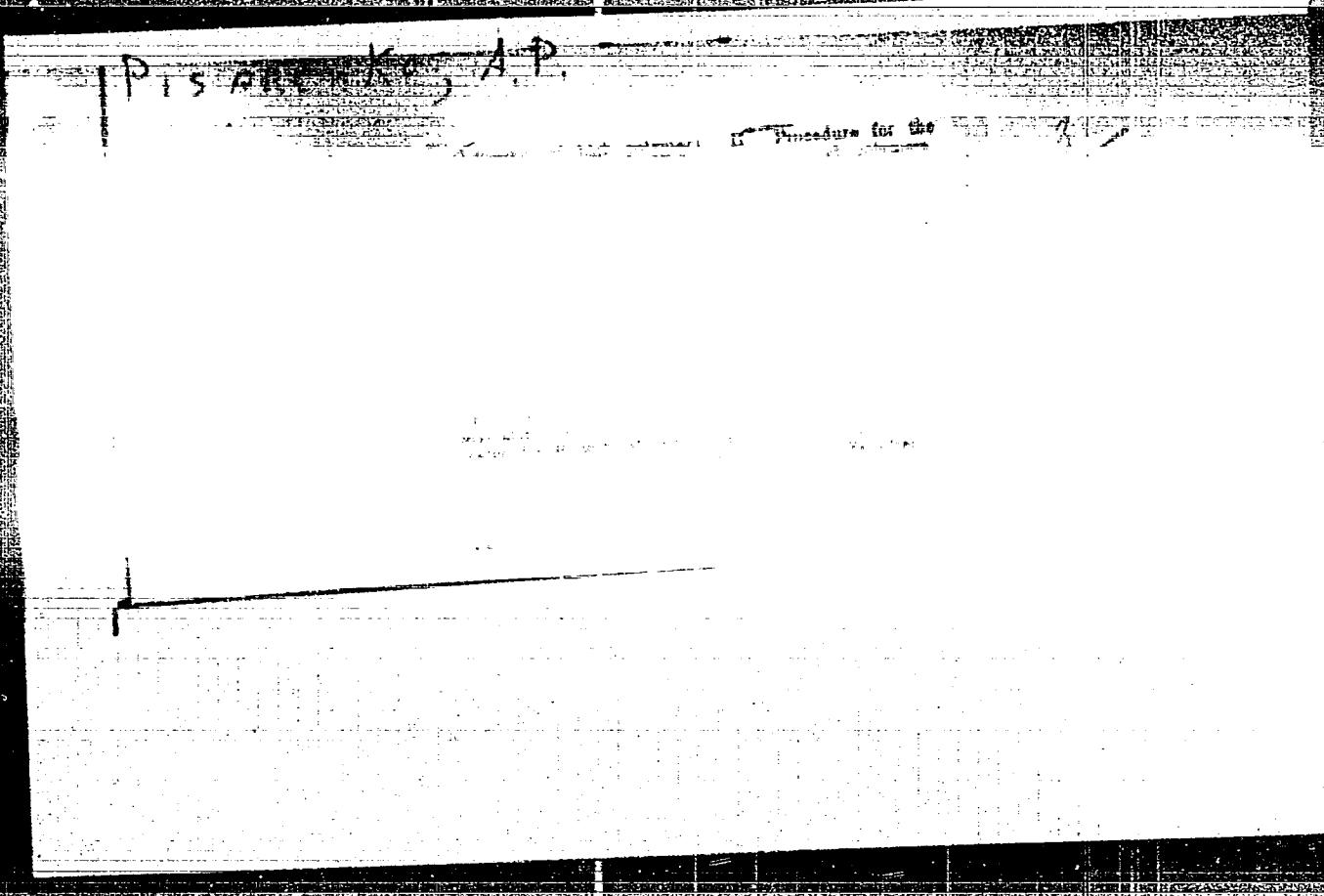
PISARENKO, A.P.

SHAPOVALOVA, A.I.; VOYUTSKIY, S.S.; PISARENKO, A.P.

Adhesion of high polymers. Part 2. Development of a method for determining
the reciprocal adhesion of high polymers. Koll. zhur. 18 no. 4:485-493 Jl-Ag
'56. (MIRA 9:10)

1.TSentral'nyy nauchno-issledovatel'skiy institut promyshlennosti zameni-
teley koshi, Moskva.
(Polymers and polymerization) (Adhesion)

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Pisarenko, A.P.

8185° New Type of Glue for Fastening Sheet Polyvinylchloride
to Metal. Novyi tip kleia dlia krepleniya lietevogo polivinil-
klorida k metallu. (Russian.) A. P. Pisarenko and R. A.
Resnikov. Legkaya Promstolennost', Moscow, Oct. 1954,
p. 23-28.
Compositions; strength, and other properties of glues. Tables,
graphs.

Pisarenko, A.P.

4

USSR.

13444* Organic Dyes for Polyvinyl Chloride. Organicheskie
krosetelli dlia polivinilchlorida. (Russian.) A. E. Pisarenko,
E. A. Libanova, and B. D. Gurdzhev. Legkaya Promstolichesk.
nost, v. 18, no. 7, July 1985, p. 25-29.
Problems of creating attractive and fast dyes; tests and struc-
tures of two dyes; phthocyanine and atraquinone pigments.
Tables, 2 ref.

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AL'BAN, M.A.; LANDA, I.M.; PISARENKO, A.P.; TUGOV, I.I.

Production of lightweight, molded, microporous footwear parts.
Leg.prom.17 no.9:13-16 S '57. (MIRA 10:12)
(Leather substitutes) (Rubber goods)

USSR Chemical Technology - Chemical Products and Their
Application - Leather. Fur. Gelatin. Tannin. Agents.
Technical Proteins.

1-2*

Abs Jour : Referat Zhur - Khimiya. N 9. 1951. 35-36

Author : Shtarkh, B.V., Pisarenko, A.P.

Inst :

Title : Electron Microscopic Study of Materials Used in the
Manufacture of Synthetic Leather.

Or & Pub : Lekkaya prom-st'. 1950. N 5. 21-30

Abstract : The work was conducted with an electron microscope of EM-3 type which permits magnification of up to 25 thousand times. Various impregnating and adhesive agents were investigated (latexes, dispersions and emulsions of high-polymers), rubbers, powders, carbon black. A drop of the dispersion was placed on a carbon-film support fastened to a brass screen. The screen was placed into camera of the electron microscope. On carrying out a comparison of a

Card : 2

USSR/Chemical Technology - Chemical Proteins and Their
Application - Leather, Fibre, Gelatin, Tanning Agents,
Technical Proteins.

Abs Jour : Ref Zhur - Khimiya, No. 1, 1951, 33-35

of synthetic latexes it was ascertained that the highest degree of particle dispersion is found in SNK-26 acrylonitrile latex. It is also noted that the latexes are polydisperse systems. The polydisperse nature of aqueous dispersions and emulsions of high-polymers has been ascertained. The investigations revealed that that particles of zinc white have an acicular crystalline structure and not a spherical as was previously assumed. Microphotographs of kaolin and sulfur show that they consist of particles of different shape

Card 2/2

LAKHTIN, A. L.; PISARENKO, A.P.

Improving the useful properties of leather substitutes with
cellulose nitrate coatings. Leg. prom. 17 no.10:29-31 O '57.
(Leather substitutes) (Nitrocellulose) (MIRA 10:12)

PISARENKO, A.P.

5155. Synthetic rubber emulsions. A. P. PISARENKO and N. I. BODINA. U.S.S.R.P.103077, nov. 29 1957. Chem. Abs., 1957, 51, 11760. Mixtures of synthetic rubber emulsions with mineral fillers are prepared by adding to the latex, aqueous solutions of silicates or stabilised hydroxides of multivalent metals and then precipitating them while simultaneously coagulating the latex.

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LAKHTIN, A.L., inzh.; PISARENKO, A.P., doktor tekhn.nauk.

Aging of nitrocellulose coatings. Leg.prom. 17 no.8:33-35
Ag '57. (MIRA №:10)
(Leather substitutes) (Nitrocellulose)

APPROVED FOR RELEASE: 07/13/2001

CIA-RDP86-00513R001341020010-9"

PISARENKO A. P.

259. Adhesion of high polymers. III. Effect of size, shape and polarity of high polymer molecules on their adhesion to Cellophane. A. S. KARWAGI, A. I. SHAPYGINA, V. V. LAVRINOV, N. D.

012 The effect of the size, shape, and polarity of macromolecules on the adhesion of a range of high polymers to Cellophane is studied with the aid of a method developed by the authors. It has been shown that the adhesion is the stronger, the lower the molecular weight of the polymer, the less the number of short side chain appendages in the macromolecule and, as a rule, the less the number in the latter of polar groups. The observed relationships have been interpreted from the standpoint of the diffusion theory of high polymer adhesion being developed by the authors. There are 12 references.

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RISARENKO, H. P.

✓ Synthetic rubber emulsion. A. P. Wissner and S. I. Rabin. U.S.P. 169,677. April 24, 1928. Stabilized synthetic rubber emulsion with mineral filler. prep. by first adding to the latex aq. solns. of silicates or stabilized hydroxides of multivalent metals and then pptg. them while simultaneously coagulating the latex.

M. Hesch

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PISARENKO, A. P.

Synthetic rubber emulsions. A. P. Blasenbrey.

וְיַעֲשֵׂה יְהוָה כָּל־אֲשֶׁר־יֹאמֵר לְךָ וְיַעֲשֵׂה כָּל־אֲשֶׁר־יֹאמֵר לְךָ

תְּמִימָנֶה אֲלֵי כָּל 3647 מִין. כָּל 15.5% בְּדַעַת

and proceeding as far as possible the first time.

prove the veracity of their statements, and to add

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PISARENKO, A. P.

Adhesion of high polymers. III. Effect of the size, shape, and polarity of the high-polymer molecules on the adhesion to cellophane. S. S. Voinovskii, A. I. Shapovalova, and A. P. Pisarenko (Central Res. Research Inst. Leather Substitutes Ind., Moscow). *Kolloid. Z. Zhar. Polym.* 19, 274-80 (1957); *cf. C.A.* 51, 10113b. — Films of polymer were deposited on cellophane by solvent evaporation and then peeled off. The resistance R to peeling was a function of the mol. wt. M had a max. at $M = 20,000$ for polyisobutylenes, and was greater for polyisoprene of $M = 20,000$ to 150,000 than for natural rubber of $M = 150,000$ to 300,000. For polybutadienes and copolymers of butadiene (I) and styrene, R increased as the no. of short side chains in the plastic decreased. For copolymers of I and $\text{CH}_2:\text{CHCN}$, R increased with the relative amt. of I. R is detd. either by cohesion or by adhesion, and the latter depends on diffusion of polymer into cellophane. J. J. Bikerman.

PISARENKO, A.P.

M Adhesion of high polymers. II. Procedure for the determination of the mutual adhesion of high polymers.
A. I. Shapovalova, G. S. Vaynshtejn, and A. P. Pisarenko.
Vysokomol. Soedin. 1956, 18, 175-83 (Engl. translat.)
See C.A. 51, 1644J.

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2 May
B.M.R.

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PISARENKO, A.P.

(4114. Properties of silicium-filled synthetic
rubbers. A. P. PISARENKO, A. P. KERSTANDOV
and N. I. KARAVANOVICH. Koch. i Resina, 1957)

14. No. 2, p. 13. Silicium-filled rubbers have high
strength, approaching that attained by adding
channel gas black to butadiene-styrene rubbers. It
is possible to prepare from these rubbers coloured
solid or microporous vulcanizates with higher
strength properties and lower consumption of
rubber. The combination of technological proper-
ties makes these rubbers attractive for the pro-
duction of tyres and technical rubber articles.
These rubbers can be used in compounds without
thermal plasticization. The advantages of these
rubbers to the national economy are stressed. They
may be used in powdered form. On the basis of
the positive results achieved with these rubbers,
SKK-80-BB and SKK-80-A-116, their full-scale pro-
duction is recommended. There are 11 references.

152D-61MD92-543186-R

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PISARENKO, A. P.

Finishing oilcloth and leather substitutes. A. P. Pisarenko and A. N. Narinetsaya. U.S.S.R. 104,213. Nov. 25, 1958. Oilcloth and leather substitutes having an oil, rubber, or polyvinylchloride base are coated with polyamide resins to improve adhesion and water-resistance of the coating. The polyamide resins are modified with *p*-butylphenyl-*tert*-BO resin.

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JOHN H. KENNEDY, JR., NEW YORK, N.Y., 1967-68, • • •

APPROVED FOR RELEASE: 07/13/2001

CIA-RDP86-00513R001341020010-9"

PISARENKO, A. P.

✓See. High temperature vulcanization of rubber
articles. A. P. PISARENKO, V. I. ALEXEENKO, and
N. S. OLEKARENKO. Lekbasy Prom., 1955, No. 9, MT
18; Transl. Cind. Lists Russ. Period., 1956, No. 83,
87.

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PISARENKO, A.P.

4116 PISARENKO, A.P. In 1958 polymer reaction
scheme of S. N. VSEK, V. A. I. KARAEV,
and A. P. PISARENKO was published in JOURNAL OF POLYMER
SCIENCE, 30, 1073; 1958, 31, 1114, 1959, 32, 1782. The
authors further presented an "infiltration theory" pro-
posed by them in previous articles this journal,
1952; abstr., 4754; 1955, abstr. 1813, 1863. According
to their theory, adhesion is accompanied by a
diffusion of the chain molecules of adhesives, etc.,
when there is a noticeable swelling of the sub-
stratum, a diffusion of the substratum molecules as
well as of the adhesive molecules, with a formation
of a combination of the type of welds, with a
gradual variation from one polymer to the other.

ACCESSION NO: AP4017166

S/0138/64/000/002/0052/0053

AUTHORS: Shvetsov, V. A.; Frenkel', R. Sh.; Pisarenko, A. P.; Zalesskaya, A. D.

TITLE: The use of native clay as raw material for the rubber industry

SOURCE: Kauchuk i rezina, no. 2, 1964, 52-53

TOPIC TAGS: rubber, vulcanized rubber, filler, clay, brown clay, kaolin, physico-mechanical property, scorching, wear, tensile strength, stretch, modulus, deformation, SKS 30 synthetic rubber, SKN 26 synthetic rubber

ABSTRACT: The present study was undertaken to find out whether the abundant brown Khvalynsk clays of the Pochtar deposit in the vicinity of the Volga Chemical Industrial Combine could be substituted for kaolin as a filler for SKS-30 and SKN-26 rubber. The brown clay contains (in %) 54.6 SiO₂, 19.1 Al₂O₃, 8.7 Fe₂O₃, 3.4 CaO, and 3.9 MgO, while kaolin contains 46.5 SiO₂, 39.5 Al₂O₃, and no Fe₂O₃, CaO, or MgO.. The specific surface of the brown clay is 56.0 m²/g as against 25.0 m²/g for kaolin. The working of the standard rubber compounds containing either brown clay or kaolin was conducted on laboratory rolls, and the physical and mechanical properties of the obtained vulcanizates evaluated by standard techniques. It was found that in plasticity and resistance to scorching both clays were practically identical.

Card 1/2

ACCESSION NO: AP4017166

but the rate of vulcanization in the presence of brown clay was enhanced, requiring at 143C only 30 minutes as against 50 for kaolin, and the strength of the obtained vulcanize was higher. However, it required nearly 70-80 parts by weight of the brown clay, as against 40 parts of kaolin, to bring about an optimal strength in the vulcanize. It was also established that the vulcanizates containing the brown clay have a higher modulus index, a lesser degree of hysteresis, and a higher endurance under multiple deformation stress than kaolin-filled vulcanizates. Orig. art. has: 1 table and 2 charts.

ASSOCIATION: Volzhskiy filial nauchno-issledovatel'skogo instituta rezinovoy promyshlennosti (Volga Branch of the Scientific Research Institute of the Rubber Industry)

SUBMITTED: 00

DATE ACQ: 23Mar64

ENCL: 00

SUB CODE: CH

NO REF Sov: 000

OTHER: 000

Card 2/2

PISARENKO, Aleksandr Pavlovich; SAFRAY, Boris Aleksandrovich; PANOVA, A.V.,
retsenzient; TCHMOZOVA, L.I., redaktor; MEDVEDEV, L.Ya., tekhnicheskiy redaktor

[Technology of the production of footwear rubber parts] Tekhnologiya
proizvodstva obuvnykh rezinovykh detalei. Moksva, Gos. nauchno-tekhn.
izd-vo Ministerstva legkoi promyschl. SSSR, 1956. 182 p. (MLRA 9:10)
(Shoe industry) (Boots and shoes, Rubber)

PISARENKO - A.P.

3
Rate of diffusion in the mutual adhesion phenomena of high-
polymer. S.S. Yermak, A.I. Shapornikova and A.P. Pisarenko
(Dokl. Akad. Nauk, SSSR, 1955, No. 1000-1005).
and contact-potential theories of the mutual adhesion of high-
polymers are discussed and rejected. Adhesion is attributed to
molecular diffusion, i.e., to mutual solution of the two polymers at
the surface of contact. This supposition is borne out by the increase
of adhesion energy as the curing temp. is raised (diffusion rate and
molecular mobility would increase with rising temp.), and by the
observation that the energy of adhesion increases (up to a certain
limit) during a long period after the join. The authors' theory
also accounts for the non-adhesion of vulcanized polymers.
N.W. Kukushkin

M.A.YOUTZ

scopies

PISARENKO, A.P.; ALEKSEYKO, V.I.; OL'SHANETSKIY, M.S.

High temperature vulcanization of rubber articles. Leg.prom. 15
no.9:18-22 S '55. (MLRA 9:1)
(Vulcanization)

VOYUTSKIY, S.S., SHAPOVALOVA, A.I., PISARENKO, A.P.

Role of diffusion in the mutual adhesion of high polymers. Dokl.
AN SSSR 105 no.5:1000-1002 D '55. (MLRA 9:3)

1. Tsentral'nyy nauchno-issledovatel'skiy institut zameniteley
kozhi. Predstavлено akademikom V.A. Karginym.
(Polymers and polymerization) (Adhesion)

PISACHENKO, A.P.

The reaction of triazine phenols with acids. V. Ya.
Kazakov, A.I. Tolstovskiy, and A. P. Pisachenko (Kiev
Institute of Organic Chem., USSR Acad. Sci., Institute of Organo-
mineral Compounds); Zh. Org. Khim. 1970, 6, 1014 (in
Russian); J. Org. Chem. 1970, 35(1), 3004. — Treatment $\text{MeCl}_2(\text{OH})$ -
 $\text{C}_6\text{H}_4\text{NNHPt}$ (I) with eq. H_2SO_4 , and with AcOH ; Et_2CO , or NaOMe in Me_2O led in each case to the decompo-
sition of the ester to MeCOCH_2Ph (II), PhNH_2 , and N_2 . It was
dissolved in dry CaH_2 , gave, in addition, to II, 1-(β -naphthalenyl)-
phenylhydrazine, which decomposed, unchanged, in H_2O , m. 73°, sol. in H_2O ,
 CH_2Cl_2 , CHCl_3 , ether, and hot Et_2O , but, in H_2O , it was
decomposed, within 1 hr, when boiled in toluene, but still not
decomposed, when heated 20 hrs, in Et_2O or CaH_2 . The follow-
ing two triazine phenols also behaved similarly: 4- MeC_6H_4 -
 $\text{C}_6\text{H}_4\text{NNH}_2/\text{C}_6\text{H}_4\text{C}_6\text{H}_3\text{NO}_2$ - ρ (III), m. 94-7° (brown
crystals); the di(ρ -nitro) isomer of III, yellow oil (dil) not
cryst. on benzene, but, like III, soluble CaH_2 , Et_2O , and
in CH_2Cl_2 , CHCl_3 , CH_3COCH_3 , $\text{CH}_3\text{CO}_2\text{Et}$, and $\text{CH}_3\text{CO}_2\text{Na}$, medium crystals, m.
93-7° (brown crystals). *gr* (2)

PISARENKO, A.P.

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MAYOUTZ
2 copies

✓ The role of diffusion in interfacial mixed adhesives.
S. S. Vorotilov, A. I. Shapovalova and A. V. Kostylev
Dokl. Akad. Nauk S.S.R. 103, 1000-011967. The authors further elaborate an adhesion theory presented by them in a no. of articles (C.A. 60, 7358b; ibid., 70024; 58,

594). According to their theory, adhesion is accompanied by a diffusion of the chain mol. of adhesives, or, when there is a noticeable swelling of the substrate, a diffusion of the substratum mol. as well as of the adhesive mol., with a formation of a combination of the type of welds, with a gradual variation from one polymer to the other.

BM/pt

PISARENKO, A.P.; SOKOLOV, S.I.

About the book of S.S.Veiutskii and B.V.Shtarkh "Physicochemical processes in the formation of films from high-polymer dispersions." Reviewed by A.P.Pisarenko, S.I.Sokolov. Koll zhur. 17 no.6:475-476
B-D '55.
(Rubber, Synthetic)(Veiutskii, S.S.)(Shtarkh, B.V.)
(MLRA 9:4)

PISARENKO, A. P.

"Physicochemical Factors in the Process of Vulcanization of Sodium-
Butadiene Rubber." Sub 28 Jun 51, Inst of Physical Chemistry, Acad Sci
USSR.

Dissertations presented for science and engineering degrees in
Moscow during 1^51.

SC: Sum. No. 480, 2 May 54

PISARENKO, A.P., professor, doktor khimicheskikh nauk; REZNIKOVA, R.A..
~~Kandidat~~ tekhnicheskikh nauk.

New type of glue for gluing polyisobutylene sheets to metal. Leg.
prom. 14 no.10:23-26 0 '54. (MLRA 7:11)
(Glue)

NARINSKAYA, A.R.; PISARENKO, A.P.; ALEKSEYENKO, V.I.

Improving the properties of polyamides used as finishing
coatings. Kozh.-obuv.prom. no.2:17-20 F '59. (MIRA 12:6)
(Resins, Synthetic) (Leather, Artificial)

L 41002-65 EPT(m)/EPT(c)/EPT(j) Po-4/Pr-4 EX
5/0031/64/000/022/B061/B064 21
ACCESSION NR: AR5005850 30

SOURCE: Prof. Dr. Khmalya, Abn. 222459

TYPE: Article, Scientific, English

TOPIC: Structure of films of SBR-30 latex filled with chalk.

CY-20 SOURCE: Nanosco-angled, tr. Vses. R.-I., Inst. plasto i gumeny, 100 ml, sb.

TOPIC: Structure of films of SBR-30 latex filled with chalk.
SUBTOPIC: Structure of films of SBR-30 latex filled with chalk, vulcanized at 130°C.
SUBTOPIC: Structure of films of SBR-30 latex filled with chalk, vulcanized at 130°C.

TRANSLATION: The structure of films of SBR-30 latex was studied by determining the kinetics of wetting and the extractability of the water-soluble substances. Films having a thickness of 0.35-0.37 mm were prepared by drying at 20°C and, in individual cases, were vulcanized for 30 minutes at 130°C. For films containing 10% (based on the weight of the polymer) ammonium caseinate, equilibrium swelling was achieved after 192 hours, compared to 130 hours for vulcanized films containing ammonium caseinate, chalk, a

softening agent and vulcanizing group, and 144 hours of non-vulcanizing group. The absolute wettability of films containing chalk was significantly lower than that of those with lamp black. This lower wettability of films containing chalk is apparently due to the interaction of chalk and ammonium caseinate with the formation of hydrophobic calcium caseinate. In order to test this hypothesis, model systems of latex with calcium carbonate were prepared. The calcium caseinate was prepared by mixing the corresponding amounts of ammonium caseinate and chemically pure CaO until formation of a homogeneous mass. The conditions under which the calcium carbonate was mixed with latex, as well as the concentration of the latex, had a marked effect on the stability of the latex stabilized with calcium caseinate. For films stabilized with calcium carbonate, complete swelling was achieved after 34 hours. The absolute wettability of latex films containing calcium caseinate was significantly lower than that of films stabilized with ammonium caseinate, regardless of the concentration of calcium caseinate. An increase in the amount of calcium caseinate led to a decrease in the wettability of the films. Addition of chalk and a softening agent to the latex led to a decrease in the content of extractable substances. These ingredients apparently facilitate the formation of a

KHOROSHAYA, Yel., KOROLEVA, K.V., SIDOROVYEV, Alex., Plastichesk.

Express method for determining the degree of acetilation of
rubber polyvinyl formal films. Kosmotekhnika, prom. 6 no. 3,
19-20 S '64.

NARINSKAYA, A.R., kand. tekhn. nauk; LISARENKO, A.P., doktor khim. nauk

Changes in the coloring of latex coatings in the interaction
of some ingredients of latex compounds. Nauch.-issl. trudy
(MIRA 18:12)
VNIIPIK no.14:69-75 '63.

ABRAMOVA, V.V., starshiy nauchnyy sotrudnik; PLOTNIKOV, I.V., kand. tekhn. nauk; FREYDGETM, K.I., mладший научный сотрудник; FISARENKO, A.P., doktor khim. nauk, prof., PAVLOV, S.A., doktor tekhn. nauk, prof.

Manufacture of artificial suede type leather without salt washout. Nauch.-issl. trudy VNIIPIK no.14-156-163 '63.
(MIRA 18:12)

GEORGIYEVA, V.S.; PISARENKO, A.P.

Effect of the conditions of gelatinization on the properties of
polychloroprene-polyvinylformaldehyd microporous films.
Kozh.-obuv.prom. 6 no.1:31-34 Ja '64. (MIRA 17:4)

AL'BAM, M.A.; PISARENKO, A.P.; LAZARYANTS, E.G.; Prinimali uchastiye:
ALADINSKAYA, I.P.; VOLKOVA, S.A.; DYUNINA, V.G.; GROMOVA, V.A.;
KOSMODEM'YANSKIY, L.V.; KOPYLOV, Ye.P.; RUKHMISTROVA, A.P.;
SHUSHKINA, Ye.N.

High-styrene rubber mixtures for the manufacture of microporous
non-shrinking rubbers. Kauch. i rez. 22 no.7:1-3 Jl '63.
(MIRA 16:8)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut plenochnykh
materialov i iskusstvennoy kozhi i Nauchno-issledovatel'skiy
institut monomerov dlya sinteticheskogo kauchuka.
(Rubber, Synthetic)

GEORGIYEVA, V.S.; PISARENKO, A.P.; SHTARKH, B.V.

Modification of polymers as a method for the production of porous structures. Dokl. AN SSSR 151 no.3:634-637 Jr. '53. 'MIRA 1c:2)

1. Predstavлено akademikom P.A.Rebindero.

(Polymers) (Porous materials)

L 29999-65 EWT(n)/EPP(s)/EPR/EWP(j)/T Po-l./Pr-l./Ps-l. W/W/KM

ACCESSION NR: AP4047675

S/0303/84/000/905/0012/0015

39

27

8

AUTHOR: Tsvetkova, N.A.; Pisarenko, A.P.

TITLE: The properties of polyesterurethane films at various temperatures

SOURCE: Lakokrasochnye materialy i ikh primeneniye, no. 5, 1984, 12-15

TOPIC: POLYESTERURETHANE RESIN, TOUYLENE DIISOCYANATE, STRESS RELAXATION, METHYLETHYLKETONE, RUBBER, POLYMER STABILITY, PLASTICITY, ARTIFICIAL LEATHER, CROSS LINKED STRUCTURE, INTERMOLECULAR BOND

RESULTS: The mechanical properties, relaxation, absorption, equilibrium modulus, number of cross links and their weights were determined for a series of polyesterurethane resins. The resins were prepared by the reaction of polyester with toluylene diisocyanate at 70 - 80°C, and were dissolved in acetone or methylethylketone for film casting. Films prepared from a resin which was synthesized at 80°C in 30 min. possessed high stability. With an increase in temperature from 20 to 70°C, these films showed a sharp decrease in stability (more than 80%). The stability of the films at 20°C was dependent on the solvent used; at 70°C this dependence was not observed. An increase in temperature on films cast

2

dependence of properties and structure on temperature was also investigated for films cast from high viscosity solutions. It was noted that an increase in temperature up to 70C decreased film stability more than 90% in comparison to the stability at 20C. The equilibrium modulus and number of bonds between the polymer macromolecules were also decreased at this temperature. Films cast from high viscosity acetone solutions had a lower stability. This information is useful in selecting the optimum conditions for producing polyesterurethane resins for artificial leather finishing. "The polyester was synthesized by L. A. Datskevich at the MKhTI im. D.I. Mendeleyeva, for which the authors express their thanks." Orig. art. has: 2 tables, 2 formulas and 3 figures.

ASSOCIATION: none

SUBMITTED: 00

ENCL: 00

SUB CODE: MT, OC

NO REF Sov: 011

OTHER: 000

Card 2/2

L 18163-63

ACCESSION NR: AP3004250

S/0138/63/000/007/0001/0003

AUTHORS: Al'bam, M. A.; Pisarenko, A. P.; Lazaryants, E. G.

45

TITLE: High-styrene rubber for nonshrinking microporous vulcanizates

SOURCE: Kauchuk i rezina, no. 7, 1963, 1-3

TOPIC TAGS: polymerization, copolymer, vulcanized rubber, shrinkage, polystyrene

ABSTRACT: The objective of the present work was the improvement of microporous rubber produced at the Kuyby*shevskiy zavod SK (Kuyby*shev Plant SK) for shoe soles. This rubber shrank excessively unless subjected to heat treatment. It was decided to replace the polystyrene component of the earlier microporous rubber by a copolymeric resin containing 85%, 90%, and 95% polystyrene. The vitrification temperature was found to increase with the percentage increase of polystyrene in the resin. A pilot plant batch of synthetic rubber with 95% polystyrene resin, calculated to contain 50% polystyrene, gave within 30 days only a 0.73% shrinkage, as against 4.64% for synthetic rubber as such. It was also observed that the use of a polymerization regulator (such as diperoxide) during the polymerization process had a detrimental effect on the shrinkage of the rubber. The extent of polystyrene polymerization as well as the temperature under

Card 1/2

L 18163-63

ACCESSION NR: AP3004250

which the process was conducted also affect the shrinkage, 100% for the former and 5C for the latter being optimal. I. P. Aladinskaya, S. A. Volkova, V. G. Dyunin, (VNIIPIK), V. A. Gromova, L. V. Kosmodem'yanskiy, E. P. Kopylov, A. P. Rokhmistrova, and Ye.N. Shushkina, (NIIMSK) participated in the work. Orig. art. has: 2 charts and 2 tables.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut plenochnykh materialov i iskusstvennoy kozhi i nauchno-issledovatel'skiy institut monomerov dlya sinteticheskogo kauchuka (All-Union Scientific Research Institute of Sheet Materials and Artificial Leather. Scientific Research Institute of Monomers for Synthetic Rubber)

SUBMITTED: QC

DATE ACQ: 21Aug63

ENCL: 00

SUB CODE: MA

NO REF SOV: 001

OTHER: 000

Card 2/2

LITNINENKO, A.G.; NOVIKOV, V.S.; PLSARENKO, A.P.

Masterbatches of carbon black rubber for shoe soles. Kozh.-obuv.
prom. 5 no.5:20-24 My '63. (MIRA 16:5)
(Boots and shoes, Rubber) (Carbon black)

AL'BAM, M.A., kand.tekhn.nauk; DYUNINA, V.G., inzh.; PISARENKO, A.P., doktor
khimicheskikh nauk, prof.

Ways of reducing the shrinkage of light-weight ~~microporous~~ ^{rubbers} (MIRA 16:3)
Isv.vys.ucheb.zav.; tekhn.leg.prom. no.1:35-44 '63. (MIRA 16:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut plenochnykh mate-
rialov i issusstvennoy khoshi (for Al'bam, Dyunina. Zauchnyy
institut sovetskoy torgovli (for Pisarenko). Rekomendovana kafedroy
khimii Zauchnogo instituta sovetskoy torgovli.
(Rubber)

L1382463 EPR/EPR(+) /EPT(c) /EPT(m) /EDS/ES(w)-2 AFPC/APDC/ASD/ESD-3/SSD
ACCUMULATED BY: AP2007002 RM/WF 8/0020/63/151/003/0634/0637 92
85

AUTHOR: Gorshkova, V. S.; Pisarevko, A. P.; Shtark, B. V.

TITLE: Modification of polymers—a technique for obtaining porous structures

SOURCE: AN SSSR. Doklady, v. 151, no. 3, 1963, 634-637

TOPIC TERM: poly(chloroprene), poly(vinyl formal), chloroprene, vinyl formal, poly(vinyl alcohol), condensation structure formation, latex, poly(chloroprene) latex, acetalization, gel, formaldehyde, acid catalyst, gelation, L-4, porous film, porous membrane, porous structure, polar polymer, nonpolar polymer, condensation network, coagulation network, synthetic leather, synthetic fiber

ABSTRACT: Poly(chloroprene-poly(vinyl formal)) porous films have been prepared by the technique of condensation-(network-)structure formation (N. V. Mikhaylov and P. A. Bobinder, Koll. zhurn., 18, no. 2, 107 (1955)). A small (unspecified) amount of poly(vinyl alcohol)(PVA) in aqueous solution was added to poly(chloroprene (KCP) latex). The PVA formed a coagulation network, and a gel resulted. Acetalization of a film of the gel with formaldehyde in the presence of an (unspecified) acid [catalyst] resulted in a stiff condensation network of poly(vinyl formal) (PVP), forming a porous structure. The critical factors

Card 1/3

L 13824-63

ACCESSION NO: AP3003862

In this process were component ratio, gelation time, and degree of dehydration prior to acetalization and acetolysis conditions. The porosity, strength, and elasticity of the porous structure can be controlled by varying gelation time. The degree of acetalization is greatly affected by the temperature and time of acetalization and by the type and concentration of acid [catalyst] used (data not given). Films based on L-4 latex and PVA and having a good microporous structure were obtained. The films demonstrated the following wide range of properties: apparent specific gravity, 0.40-1.20 g/cm³; (water) vapor permeability, 0.05-710 mg/cm² hr; water absorption, 2 hr, 3-50%; 24 hr, 3-60%; tensile strength, 10-80 kg/cm²; and elongation-at-break, 100-900%. Study of the solubility of the films in water, glacial acetic acid, and chloroform revealed that 50% of the PVP is not extracted by the acid, even at 61.3-7°C, suggesting that the network structure is due to bonds stronger than Van der Waals forces. IR spectra were compared for PVP and for PCP-PVF gels swelled in chloroform after prolonged high-temperature extraction in glacial acetic acid. A substantial number of bands characteristic of PVP were also observed in the gels, confirming the feasibility of modifying PCP by formation of a PVP phase in the system. The equilibrium modulus of PCP-PVF films (13.9 kg/cm²) sharply exceeded that of PCP-PVA (4.90) or PCP (2.15) films. It is stressed that in the manufacture of synthetic leather and fibers, the condensation-[network]-structure

Card 2/3

L 13824-63

ACCESSION NO: AF2007362

fusion techniques makes it possible not only to produce porous structures, but also to combine polar and nonpolar polymers and to vary properties according to the desired characteristics of the end product. "The authors express their profound gratitude to P. A. Rebinder, A. B. Taubman, I. N. Vlodavets, Ya. N. Yushko, M. M. Bernshteyn, and M. P. Lifschits for giving a number of valuable opinions during the discussion of the results." Orig. art. has: 2 figures and 2 tables.

ASSOCIATION: none

SUBJECT: 087663

DATE ACQ: 15Aug63

EVAL: 00

SER CODE: CI, MA

NO REP Sov: 007

OTHER: 001

Card 3/3

L 13824-63 EPR/EPR(+) /EPF(c)/EWT(w)/EDS/ES(v)-2 ATTC/APGC/ASD/ESD-3/SSD
Po-4/Po-4/Pr-4/Pr-4 - RM/WW
ACCIDENTAL NO: AP200382 8/0020/63/151/003/0634/0637 92

AUTHOR: Gorobtseva, V. S.; Misarenko, A. P.; Shternin, B. V. 85

TYPE: Modification of polymers—a technique for obtaining porous structures

SOURCE: AN SSSR. Doklady*, v. 151, no. 3, 1963, 634-637

TOPIC CODE: polychloroprene, poly(vinyl formal), chloroprene, vinyl formal, poly(vinyl alcohol), condensation structure formation, latex, polychloroprene latex, acetalization, gel, formaldehyde, acid catalyst, gelation, L-4, porous film, porous membrane, porous structure, polar polymer, nonpolar polymer, condensation network, coagulation network, synthetic leather, synthetic fiber

ABSTRACT: Polychloroprene-poly(vinyl formal) porous films have been prepared by the technique of condensation-[network]-structure formation (N. V. Milkaylov and P. A. Rebindar, Koll. zhurn., 18, no. 2, 107 (1955)). A small [unspecified] amount of poly(vinyl alcohol) (PVA) in aqueous solution was added to polychloroprene (KCP) latex. The PVA formed a coagulation network, and a gel resulted. Acetalization of a film of the gel with formaldehyde in the presence of an [unspecified] acid [catalyst] resulted in a stiff condensation network of poly(vinyl formal) (PVF), forming a porous structure. The critical factors

Card 1/3

L 13824-63

ACCESSION NO: AP3009062

In this process were component ratio, gelation time, and degree of dehydration prior to acetalization and acetalization conditions. The porosity, strength, and elasticity of the porous structure can be controlled by varying gelation time. The degree of acetalization is greatly affected by the temperature and time of acetalization and by the type and concentration of acid [catalyst] used [data not given]. Films based on L-4 latex and PVA and having a good microporous structure were obtained. The films demonstrated the following wide range of properties: apparent specific gravity, 0.40-1.20 g/cm³; (water) vapor permeability, 0.05-710 mg/cm² hr; water absorption, 2 hr, 3-50%; 24 hr, 3-60%; tensile strength, 10-80 kg/cm²; and elongation-at-break, 100-900%. Study of the solubility of the films in water, glacial acetic acid, and chloroform revealed that 90% of the PVP is not extracted by the acid, even at 61.2-70C, suggesting that the network structure is due to bonds stronger than Van der Waals forces. IR spectra were compared for PVP and for PCP-PVP gels swelled in chloroform after prolonged high-temperature extraction in glacial acetic acid. A substantial number of bonds characteristic of PVP were also observed in the gels, confirming the feasibility of modifying PVP by formation of a PVP phase in the system. The equilibrium modulus of PCP-PVP films (13.9 kg/cm²) sharply exceeded that of PCP-PVA (4.90) or PCP (2.15) films. It is stressed that in the manufacture of synthetic leather and fibers, the conformation-[network]-structure

Card 2/3

L 13824-63

ACCESSION NR: AF9003862

Formation techniques makes it possible not only to produce porous structures, but also to combine polar and nonpolar polymers and to vary properties according to the desired characteristics of the end product. "The authors express their profound gratitude to P. A. Rebinder, A. B. Taikman, I. N. Vlodavets, Ya. M. Yabko, M. M. Bernahteyn, and M. P. Litovchenko for giving a number of valuable opinions during the discussion of the results." Orig. art. has: 2 figures and 2 tables.

6

ASSOCIATION: none

SUBJECTD: 007063

DATE ACQ: 15Aug63

INCL: 00

SUB CODE: CH, MA

NO REF Sov: 007

OTHER: 001

Card 3/3

LITVINENKO, A.G.; NOVIKOV, V.S.; PISARENKO, A.P.

Carbon black and rubber mixture batches in sheet form for footwear
soles. Kozh.-obuv.prom. 5 no.2:22-25 P '63. (MIRA 16:5)
(Leather, Artificial)

SHTARIN, N.V., CHIKHALYA, V.A., PIKREK, A.P.

aff. - on thermal treatments on the dispersity of synthetic latexes
Kaz. Polym. Inst. 7 8-10 Jl 1960

1. Vsesoyuznyy nauchno-issledovatel'skiy institut plastyrov
materialov i ekspresivnyy proyektskiy
(Rubber, Synthetic.)

KHOROSHAYA, Ye.S.; KOVRIGINA, G.I.; NARINSKAYA, A.R.; PISARENKO, A.P.

Rapid sulfite micromethod for determining the degree of latex film
vulcanization. Kauch. i rez. 20 no.12:40-42 D '61.
(MIRA 15:1)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut plemochnykh
materialov i iakusstvennoy kozhi.
(Latex) (Vulcanization)

"APPROVED FOR RELEASE: 07/13/2001 CIA-RDP86-00513R001341020010-9

APPROVED FOR RELEASE: 07/13/2001 CIA-RDP86-00513R001341020010-9"

PISARENKO, Aleksandr Pavlovich, prof.; POSPELOVA, Kseniya Aleksandrovna,
dots.; YAKOVLEV, Alexei Georgiyevich, dots.; VOYUTSKIY, S.S.,
prof., retsenzent; NAZAROV, V.I., prof., retsenzent; TAULMAN, S.S.,
prof., retsenzent; BARABOYM, N.K., prof., retsenzent; STUKOVIN,
I.D., red. izu-v.; YEZHOOVA, L.L., tekhn. red.

[Course in colloid chemistry] Kurs kolloidnoi khimii. Moskva,
Gos.izd-vo "Vysshiaia shkola," 1961. 241 p. (MIR 14:12)
(Colloids)

SHVETSOV, V.A.; NOVIKOV, A.S.; PISARKENKO, A.P.

Properties of reinforced aluminate-nitrile rubbers. Kauch.i rez.
19 no.4:12-17 Ap '60. (MIRA 13:12)

1. Nauchno-issledovatel'skiy institut rezinovoy promyshlennosti
i Vsesoyuznyy nauchno-issledovatel'skiy institut plenochnykh material-
ov i iskusstvennoy kozhi.
(Rubber, Synthetic)

SHVETSOV, V.A.; PISARENKO, A.P.; NOVIKOV, A.S.

Nature of the bonding in the system rubber - filler. Koll. shur.
22 no. 6:743-747 E-D '60. (MIRA 13:12)

1. Nauchno-issledovatel'skiy institut resinovoy promyshlennosti,
Moskva.
(Alumina) (Rubber, Synthetic) (Fillers (In paper, paint, etc.))

SHVETSOV, V.A.; NOVIKOV, A.S.; PISARENKO, A.P.

Study of the structure of reinforced butadiene-nitrile
rubber vulcanizates by the elongation method. Vysokom.
soed. 2 no. 11:1608-1612 N '60. (MIR 13:11)

1. "Nauchno-issledovatel'skiy institut rezinovoy promyshlennosti.
(Rubber)

87768

112210
15.9200

2109.1109 R129

S, CEC/MC/OM/OC/OC-10
BC11/BC66

AUTHORS: Shvetsov, V. A., Pisarenko, A. P., and Nikulin, V. A. S.

TITLE: Problem of Investigating the Bond Character in the System
Rubber - Filler

PERIODICAL: Kolloidnyy zhurnal, 1950, Vol. 12, No. 1, p. 74-78

TEXT: The authors applied the method devised by B. Degadkin and co-workers (Refs. 1 and 2) to study the dispersion of calcium silicate in aluminum hydroxide in rubber mixtures and in vulcanized rubber. The present paper reports the results with respect to aluminum hydroxide. The partial or complete extraction of aluminum hydroxide from the rubber by boiling in weak NaOH solutions was shown to be possible. The optimum extraction time (2 hours) was found to correspond to the optimal filler of 60 parts by weight. In this case a maximum development of rubber occurs with a markedly pronounced entanglement phase of the filler, which facilitates the penetration of the solvent into the rubber. With poor filling, the chain structure of the filler is less pronounced, and extraction is more time-consuming. This rule also holds if rapid thermal

Card 1/2

87768

Problem of Investigating the State Changes
in the System Rubber - Filler

rubbers (binary system rubber + filler). The extraction in this case, however, proceeds more slowly than in rubbers that have a higher degree of crosslinking, which is due to a denser structure and a higher content of rubber in the bonds. On incorporation of aluminum hydroxide during the vulcanization, compact rubbers were obtained that on incorporation of the filler into latex. The filler incorporated during vulcanization was found not to form stable structures. The authors further studied the effect of stearin on the properties of the resultant rubber. The extraction of aluminum hydroxide from medium in the presence of stearin was found to be accelerated by the introduction of stearin. In vulcanized rubbers the aluminum hydroxide extraction takes place in the same way as in non-vulcanized systems. The comparative study of the easy extraction of aluminum hydroxide from rubber mixtures and vulcanizates indicates that prevalent physical adsorption of the hydroxide on the rubber indicates that predominantly physical adsorption of the hydroxide on the rubber. Stable chemical bonds are either not formed at all or to a very low extent. After incorporation of aluminum hydroxide into the rubber in its state and after subsequent extraction of this filler the rubber retains properties which differ from the original ones. The mixture contains 10% of the basis of extracted (KH-2F SKN-2F) rubber gave much better vulcanizates.

Card 2/2

87768

Problem of Increasing Strength of
In the System Rubber + F.

3. Strength of 100%
Rubber

rubber + 10% mixture of SKN-4. Rubber + 10% mixture of SKN-4 has higher strength than the former case, i.e., 4 kg/cm² relatively to 3.5 kg/cm². The reason in the latter case is 4 kg/cm² relatively to 3.5 kg/cm². The reason is that the rubber is treated at 100°C. The higher strength of the vulcanizate is obtained from rubber which filled carbon black uniformly. It is known that structure formation takes place in the polymer chain during vulcanization. The formation of direct bonds between the polymer chain and carbon black is heat treatment of the system rubber + filler. There are literature data and references: "Svirskii, G. I. and V. S. Pashin."

ASSOCIATION National Research Institute of Synthetic Rubber
M. Svirskii (Scientist), R. V. Kuznetsov, I. V. Rukavishnikov,
V. A. Kostylev, M. S. Slobodchikov

SUBMITTED N. A. Slobodchikova

Card 1/1

11/20/11
ATTORNEY

Study of the Soviet Union's

TITLE

Study of the Soviet Union's
Economic System

PERIODICAL

U.S. News & World Report

TEXT: The following was written by a member of the Soviet Economic Institute. It was written in 1958 and was published in the Soviet magazine "Ekonomiceskaya zhurnalistika". It is a study of the Soviet economic system. It is a good source of information about the Soviet economy.

25-11-1

S. 1. *Constitutive role of the arachidonic acid in the
cellular regulation of the phosphatase*

S. 2. *Effect*

S. 3. *Arachidonic acid stimulates the phosphatase activity in
the membrane fractions of the rat brain cortex and the
rat liver. The effect is dose-dependent and reversible.*

Q.L1 1-7

Saint Lucia - St. Lucia - The Islands
of the West Indies - The Caribbean
St. Lucia

Many values are given for the same parameter, e.g., the refractive index of ordinary light.

卷之三

Study of the Structure of Various Types of
Pearl and Butadiene-Nitrile Rubber
Skins

Calorimetric methods were used to determine

the ΔH_f° of the polymerization of

Styrene

Toluene

Tetralin

Compounds of rubber

Styrene by polymerization

Natural rubber with 10% styrene

SKN-2 with 28% styrene

SKN-3 with 10% styrene

SKN-4 with 28% styrene

SKN-5 with 10% styrene

SKN-6 with 10% styrene

SKN-7 with 10% styrene

SKN-8 with 10% styrene

SKN-9 with 10% styrene

SKN-10 with 10% styrene

SKN-11 with 10% styrene

SKN-12 with 10% styrene

SKN-13 with 10% styrene

SKN-14 with 10% styrene

SKN-15 with 10% styrene

SKN-16 with 10% styrene

SKN-17 with 10% styrene

SKN-18 with 10% styrene

SKN-19 with 10% styrene

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85411

Study of the Structure of Vulcanizates
Produced by Different Rubbers by
S. N. Slepcev

S/ DC/EO/002/C 1/1 /C
PCO4/3042

rubber with channel black or aluminum hydroxide. Vulcanizates produced by natural rubber filled with aluminum hydroxide are more crosslinked at a higher degree than the same rubbers filled with barium sulfate. Natural rubber has more primary bonds while SKN 26 has a few more channel black but especially with aluminum hydroxide has a few more secondary bonds. The structure of the vulcanizates does not differ very much. The author of the article is A. P. Slepcev mentioned. There are 6 figures in the article and it was prepared in Soviet, US, British and German.

ASSOCIATION: Nauchno-issledovatel'skii vystavki i laboratoriya
pramyshlennosti "Sverdlovsk Research Institute
of the Rubber Industry".

SUBMITTED February 2, 1960

Card 1/2

SHVETSOV, V.A., PISARENKO, A.P., SHTARKH, D.V., NOVIKOV, A.S.

Electron microscopic study of the structures of reinforced
rubbers. Koll. zhur. 22 no. 2:233-236 Mr-Ap '60.
(MIR 13:8)

1. Nauchno-issledovatel'skiy institut plenochnykh materialov
i iskusstvennoy kozhi i Nauchno-issledovatel'skiy institut
rezinovoy promyshlennosti, Moskva.
(Rubber--Research)

83837

159200 also 1109

S/138/60/000/004/001/008
A05/A029

11.2220

AUTHORS: Shvetsov, V A., Novikov, A S., Pisarenko, A P.

TITLE: The Properties of Filled Aluminate-Nitrile Rubbers 15

PERIODICAL: Kauchuk i Rezina, 1960, No. 4, pp 12 - 17

TEXT: The results of the development of a method for producing nitrile rubbers filled with aluminum hydroxide in the latex (called aluminate-nitrile rubber) are given. With this method it is possible to produce vulcanizates with high physico-mechanical properties. No complex apparatus is necessary and the aluminate-nitrile rubbers have a higher mechanical resistance than the silicate-nitrile rubbers. The CKH-18 (SKN-18) and the CKH-26 (SKN-26) latexes were used in the production of the aluminate-nitrile rubber. Vulcanizates produced from aluminate-nitrile rubber have a high elasticity. The specific elongation in SKN-26 reaches 1,000 - 1100%. Rubbers with aluminum hydroxide obtained from the latex, as well as that with introduction on the rollers have an elevated residual elongation and a high resistance to wear and tear. The resistance to repeated bending and crack growth is higher in vulcanizates with aluminum hydroxide introduced into the latex

Card 1/4 X

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S/138/60/000/004/00¹/00⁴
A05¹/A029

The Properties of Filled Aluminate Nitrile Rubbers

compared to introduction on the rollers only. In order to study the resistance of the rubber to thermal effect, samples of various vulcanizates were subjected to aging in a thermostat at temperatures beginning with 12 hours and lasting up to 10 days. The best results showed rubbers with aluminum hydroxide introduced on the rollers. Aluminate nitrile rubbers can be stored for a long time without changing their properties. The authors attempted to discover why the filler obtained in the latex has a higher strengthening power as compared to the filler produced separately and introduced into the rubber on the rollers. They also tried to determine the comparative properties of mineral fillers of the potassium silicate and aluminum hydroxide types, e.g., gaseous channel carbon black. The structuralizing role played by the fillers was investigated experimentally and found to correspond favorably with previous data (Refs. 2 - 5). The experiments also showed that the high strengthening ability of calcium silicate and aluminum hydroxide obtained in the latex can be explained by the high dispersion of the particles of the filler and good distribution of the filler in the rubber mass as well as the absence of aggregation of the particles, formation of chain structures by the filler and the ability of the filler to form a bond.

Card 2/4

83837

S/138/60, 090 004 1011 1-6
AC51/A029

The Properties of Filled Aluminate-Nitrile Rubbers

of the adsorption type with the rubber. The experimental data also proved that the presence of the SO₄ group in the molecule does not affect the strengthening ability of aluminum hydroxide produced in the latex, contrary to other opinions. The SO₄ group can have an effect on the crystallization process which takes place when the filler is produced outside of the latex and, therefore, on the structure and dispersion of the filler and, thus, in directly on the strengthening ability of the latter. The method developed for producing aluminate-nitrile rubbers has great significance for the Soviet Rubber Industry, since it lowers the energy consumption, the time needed to produce the mixtures and improves the productivity of the mixing apparatus. The rubber produced in the form of chunks enables one to automate the weighing and loading processes during mixing. It also enlarges the assortment of the different rubbers used in the manufacturing of oil resistant and thermo-resistant rubber products. Finally, this method replaces the use of starch carbon black, since the aluminate-nitrile rubber yields vulcanizates similar in their properties to that of the vulcanizates on a gaseous carbon black base. There are 6 tables, 1 figure and 10 Soviet references.

Card 3/4

83837

S/38/60/000 104 001/308
A051/A029

The Properties of Filled Aluminate-Nitrile Rubbers

ASSOCIATION: Nauchno-issledovatel'skiy institut rezinovoy promyshlennosti
i Vsesoyuznyy nauchno-issledovatel'skiy institut plenochnykh
materialov i iskusstvennoy kozhi (Scientific Research Insti-
tute of the Rubber Industry and All Union Scientific Research
Institute of Film Materials and Synthetic Leather) ✓

Card 4/4

PISARENKO, A.P.; SHAPOVALOVA, A.I.; VOYUTSKIY, S.S.

Adhesion of high polymers. Part 4: Effect of mechanical factors on
high polymer adhesion. Koll.zhur. 22 no.1:57-62 Ja-P '60.
(MIRA 13:6)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennoy
koshi, Moskva.
(Adhesion) (Polymers)

S/158, 59, 60, 61, 62, 63, 64

AUTHORS: Shvetsov, V. A., Bisarenko, A. P., Novikov, A. S.TITLE: An Investigation Into the Properties of Filled Nitrile Rubbers
Communication 1. The Properties of Filled Silicate-Nitrile
Rubbers 15

PERIODICAL: Kauchuk i Rezina, 1959, No. 12, pp. 4-8

TEXT: At present two types of powdered silica gel are manufactured in the Soviet Union, viz. soft and hard silica gel, imparting different properties to the rubber. It is further stated that powdered silica gel is irreplaceable as an accelerator in the production of colored rubber based on synthetic rubbers with high mechanical indices and has many advantages over the carbon blacks. One of the most popular types of silica gel is aerosil, which is just as active as any carbon black. It is pointed out that the Soviet rubber industry lacks sufficient quantities of the silica gel accelerators and the available types have some serious disadvantages due to the backward production methods used. Their quality is not homogeneous. The necessity of producing filled rubbers based on synthetic raw material by some other means is pointed out. A short survey is given

Card 1/4

S-387, 300-1000

An Investigation Into the Properties of Filled Nitrile Rubbers.
Communication 1. The Properties of Filled Silicate-Nitrile Rubbers

of the methods recently used for this purpose. The All-Union Scientific Research Institute of Film Materials and Artificial Leather (VNIPIK) developed in 1951-1953 a method for the production of filled butadiene-styrene rubbers, using silicates of various metals obtained in the latex as fillers. The CKH-18 (SKN-18), CKH-26 (SKN-26) and CKH-40 (SKN-40) type butadiene-nitrile rubbers are used in the rubber industry for the production of oil-resistant rubber. The authors were particularly interested in determining the possibilities of producing oil-resistant and heat-resistant butadiene-nitrile rubbers, filled with silicate fillers during the latex stage. These rubbers were named silicate-nitrile rubbers. It was shown that the strength of the rubber increases considerably when the filler is introduced in the latex stage, and much less so, when introduced on the rollers. This is true even for small quantities of the filler such as 20 weight parts of filler to 100 weight parts of the rubber. For greater amounts of filler, e.g. 60 weight parts of filler to 100 weight parts of rubber the relative elongation is 575-674%. The high structuralizing properties of the silicate fillers obtained in the latex can be seen from the hardness determination according to Defoe. The mechanical properties of the

Card 2/4